

Evaluation of amide and C28 phases for separation of hydrophilic compounds

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Abstract

Hydrophilic Interaction liquid Chromatography (HILIC) proposed by Alpert in 1990 has been applied for analysis of hydrophilic compounds. Amide, diol, polyol, bare silica, ion exchange and zwitter ion phases have been used as a hydrophilic stationary phase along with an organic solvent rich mobile phase. A polar group embedded C18 or a long alkyl chain phase such as C30 or C28 also have been used to separate hydrophilic compounds without change in retention using an aqueous mobile phase on a reversed-phase mode. The reason why these reversed-phases can show no change in retention under an aqueous condition is that a low contact angle of water on the surface of the pore of these reversed-phase packing materials makes an aqueous mobile phase keep in the pore because pressure yielded by capillarity is less atmospheric pressure, so that retention doesn't change. Both HILIC stationary phases and reversed-phases have completely opposite characteristics each other. Therefore both HILIC and reversed-phase modes are useful for separation of hydrophilic compounds. It is important to understand separation behavior of each mode. In this study, an amide column and a C28 column were compared and evaluated to separate hydrophilic compounds. SunShell HILIC-Amide and Sunniest RP-AQUA (C28) and SunShell RP-AQUA (C28) were used to separate nucleobases, amino acids and hydrophilic vitamins. When nucleobases were separated on HILIC and reversed-phase modes using an amide column and a C28 column, each elution order of samples is said to be opposite. Only uracil, however, showed a specific elution. It was considered that the polarity of uracil under an organic solvent rich condition was different from that on water rich condition to be due to keto-enol tautomerization. LC/MS analysis of amino acids was achieved using C28 column and a mobile phase added 5 mM heptafluorobutyric acid under gradient elution conditions.

Collapse or Depermeating

C18 phases exhibit decreased and poorly reproducible retention under more than 98% aqueous conditions. This problem traditionally has been explained as being the result of ligand collapse. Nagae¹⁻³ ascertained, however, that the mobile phase was being expelled from the pores of the packing material.

When the surface of packing materials isn't wet by water, water used as a mobile phase expels from the pore of the packing material by capillarity. This is a reason why reproducibility in retention is low under 100% aqueous conditions. Reversely pressure around the pore of the packing material makes water permeate into the pore of the packing material to overcome a force worked by capillarity.

1) N. Nagae, T. Enami and S. Doshi, LC/GC North America October 2002.
2) T. Enami and N. Nagae, American Laboratory October 2004.
3) T. Enami and N. Nagae, BUNSEKI KAGAKU, 53 (2004) 1309.

What does "Dewetting" mean?

A surface state changes from wetting to un-wetting? The surface of C18 is always un-wetting even if water exists in the pore, so that expression of "dewetting" is not scientific. Depermeating is a scientific expression!

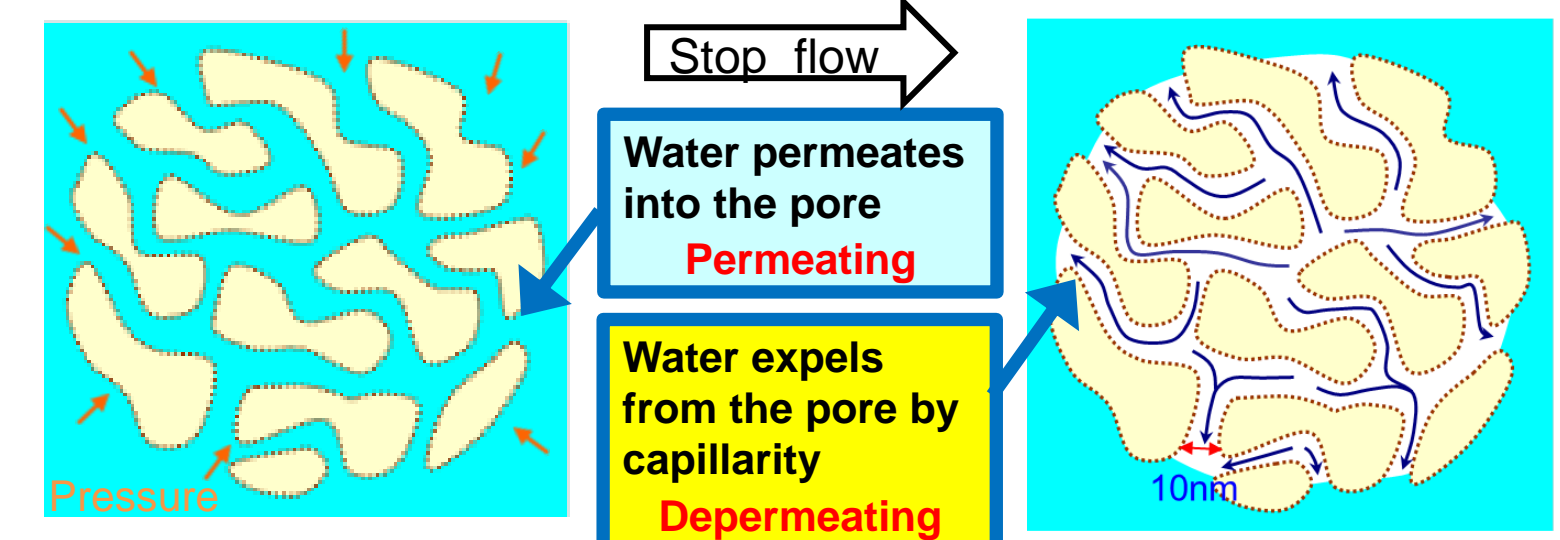


Figure. 6 Schematic diagram of C18 particle

Formula of Capillarity:
 $h = \frac{2\gamma \cos\theta}{r\rho g}$
 γ : Surface tension
 ρ : Density of liquid

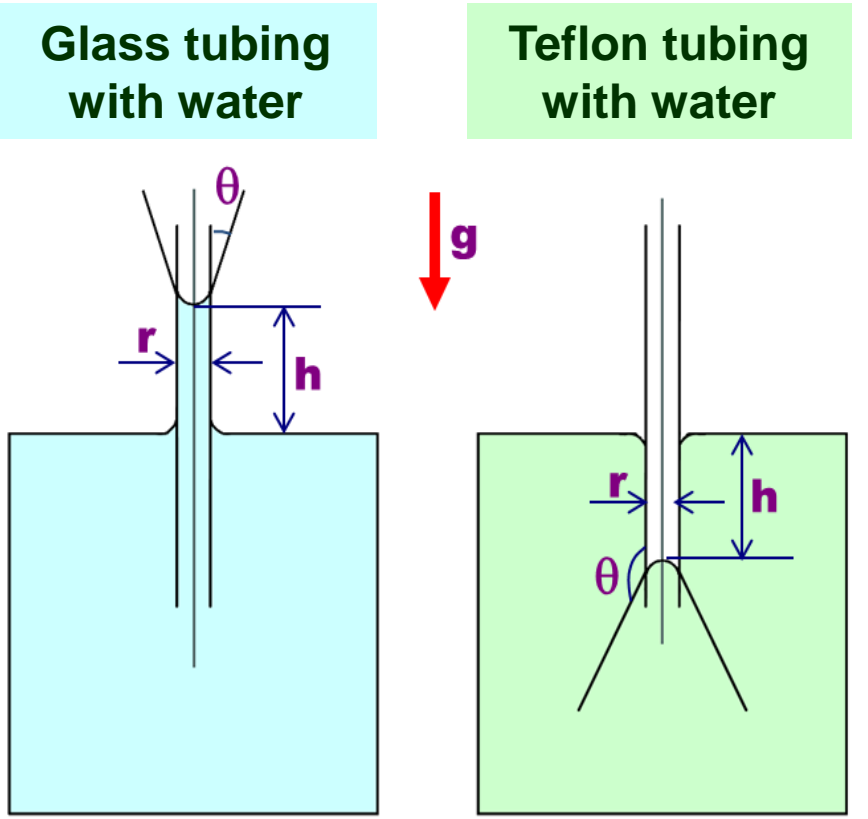
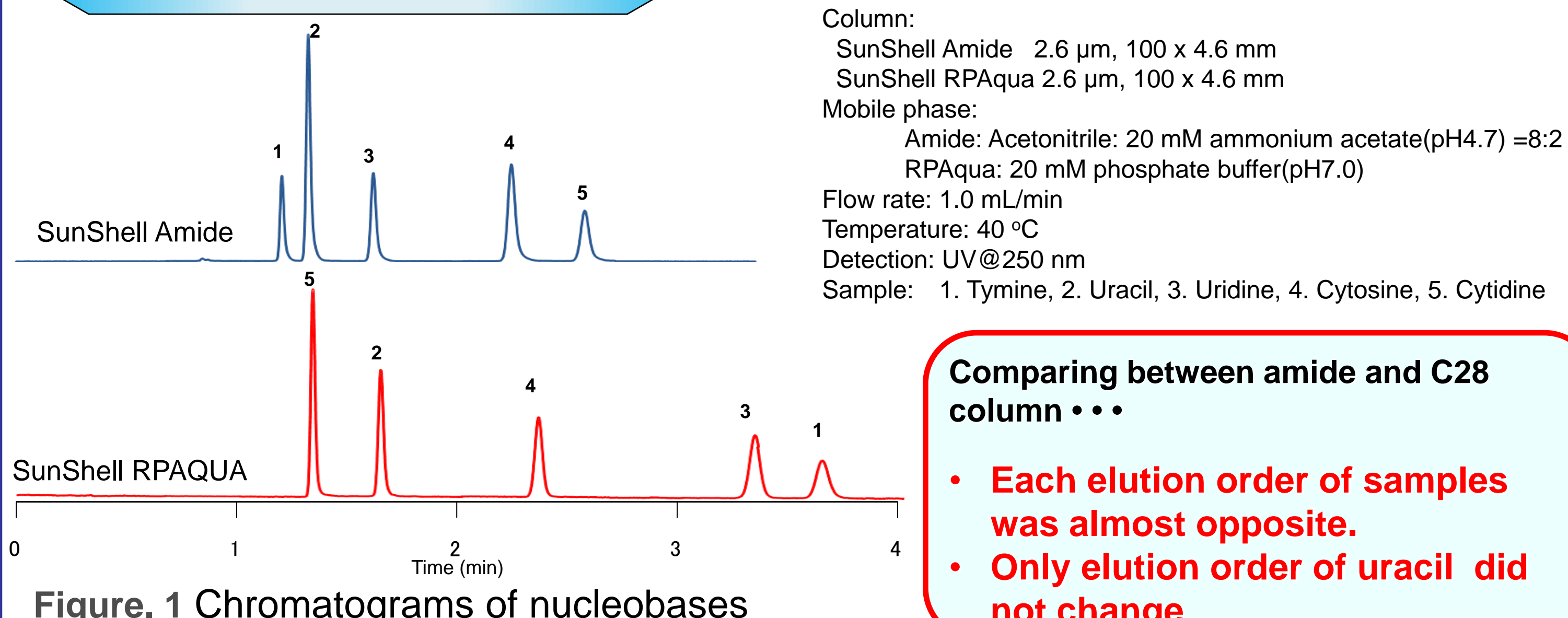
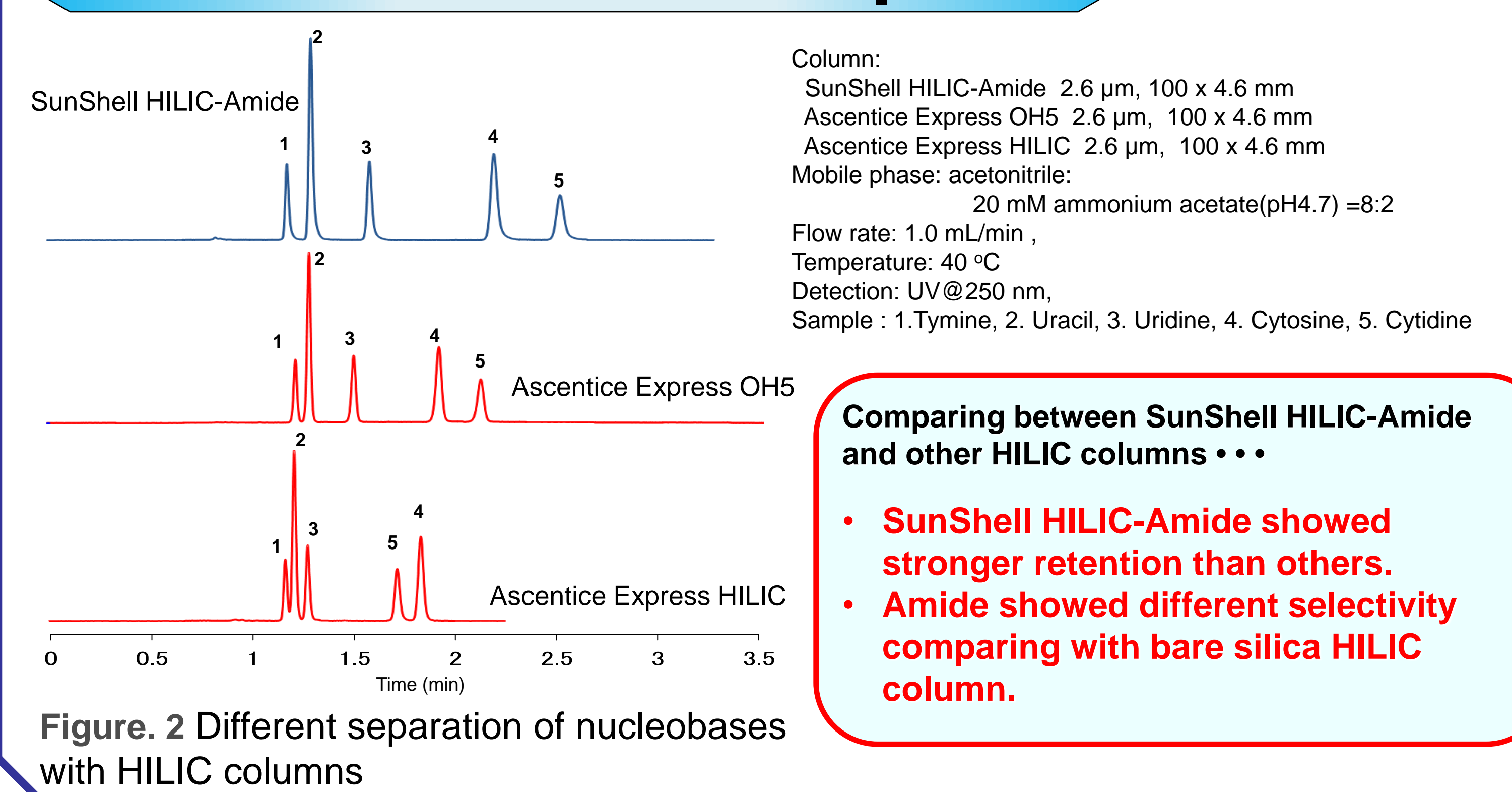


Figure. 7 Schematic of capillarity

Amide vs C28



Amide vs other HILIC phase



Applications of HILIC

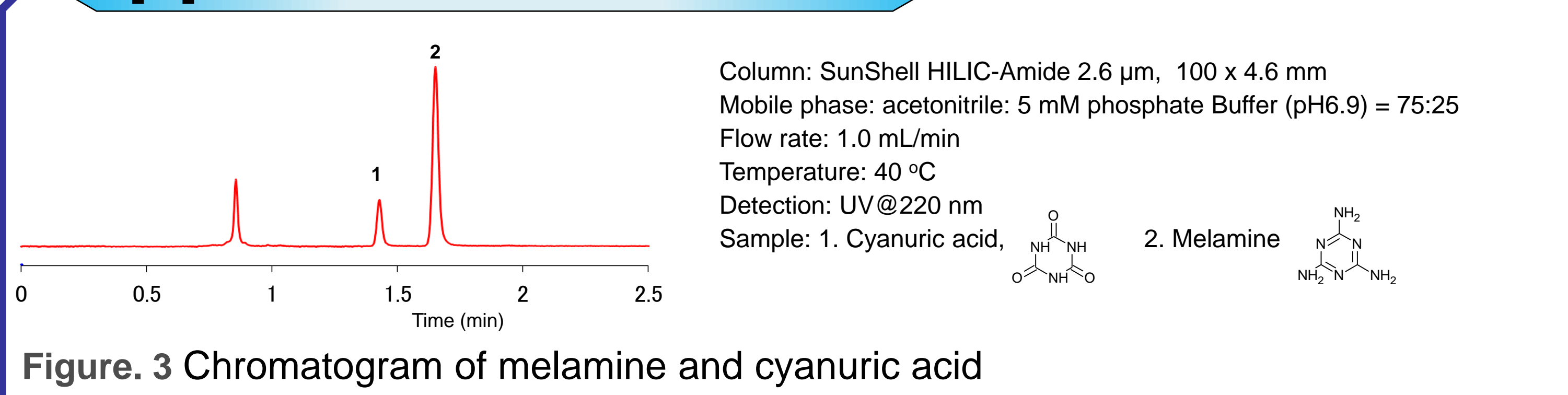


Figure. 3 Chromatogram of melamine and cyanuric acid

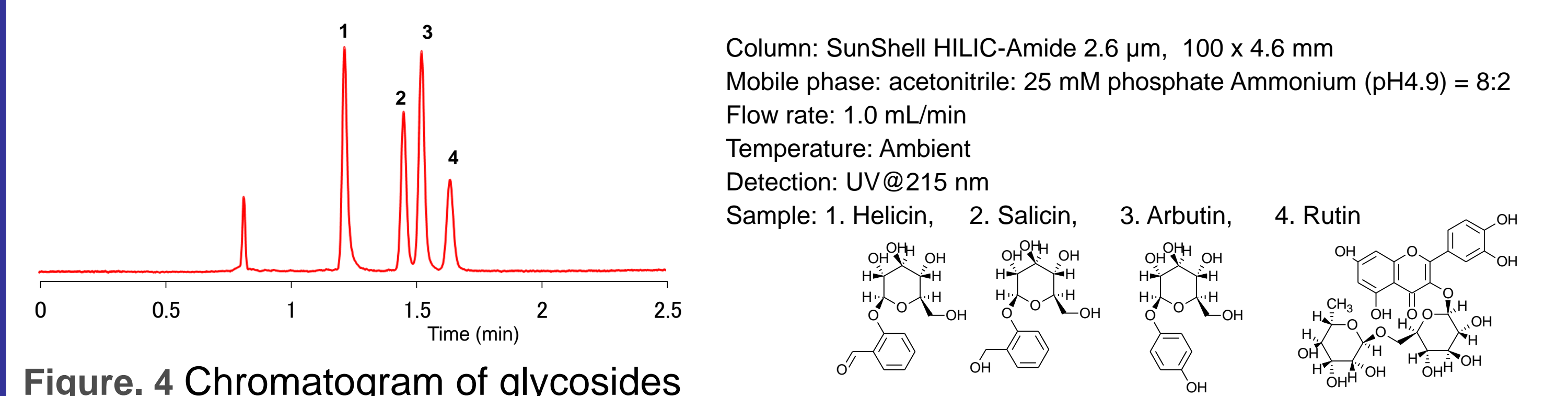


Figure. 4 Chromatogram of glycosides

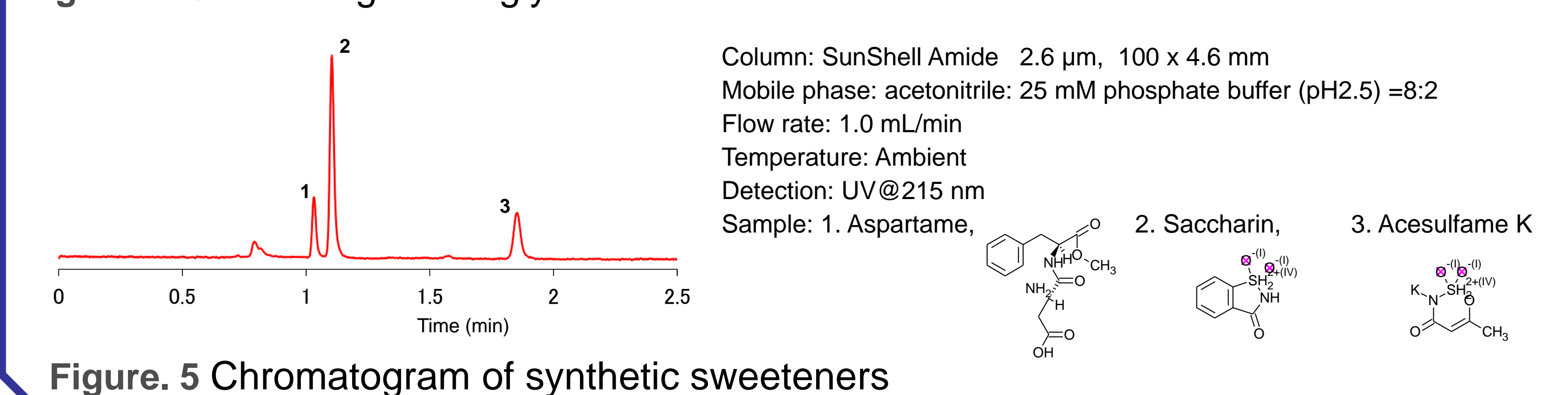


Figure. 5 Chromatogram of synthetic sweeteners

Conclusions

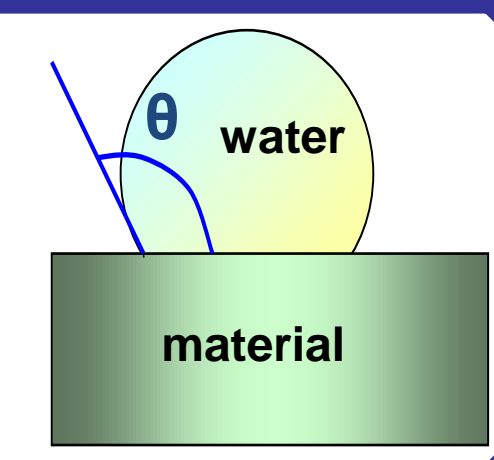
- Nucleobases were separated using an amide column and a C28 column, each elution order of samples is said to be opposite.
- Only uracil showed a specific elution. It was considered that the polarity of uracil under an organic solvent rich condition was different from that on water rich condition to be due to keto-enol tautomerization.
- LC/MS analysis of amino acids was achieved using C28 column and a mobile phase added 5 mM heptafluorobutyric acid under gradient elution conditions.
- Both amide and C28 column were useful for analysis of hydrophilic compounds

Repellency and Hydrophobicity

Repellency

Water-shedding property

- Repellency is expressed as a contact angle of water on a material.
- The larger a contact angle, the stronger repellency, if the contact angle is more than 90 degree.



Repellency \propto Hydrophobicity

- Repellency and hydrophobicity are independent each other.
- Those two parameters are out of proportional.
- When hydrophobicity is high, it doesn't mean that repellency is always high.
- Capillarity depends on a contact degree.

Hydrophobicity

Difficult to mixing with water

- Hydrophobicity is expressed as the ratio of concentrations of a compound between water and n-octanol using a mixture of both solvents.
- This value is well known as LogPow.

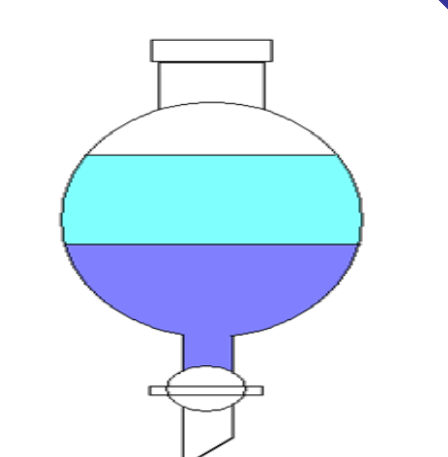


Table. 1 Physical property of each compounds

	Trifluoromethane	Octadecane (C18)	Octane (C8)	Octacosane (C28)
Contact angle(θ)	120°	126°	140°	108°
Partition coefficient (LogP)	0.64	9.18	5.18	14.09
Solubility(mg/L)	4090	0.006	0.66	8.84×10^{-10}

T. Enami and N. Nagae, BUNSEKI KAGAKU, 53 (2004) 1309.

Stability of C28

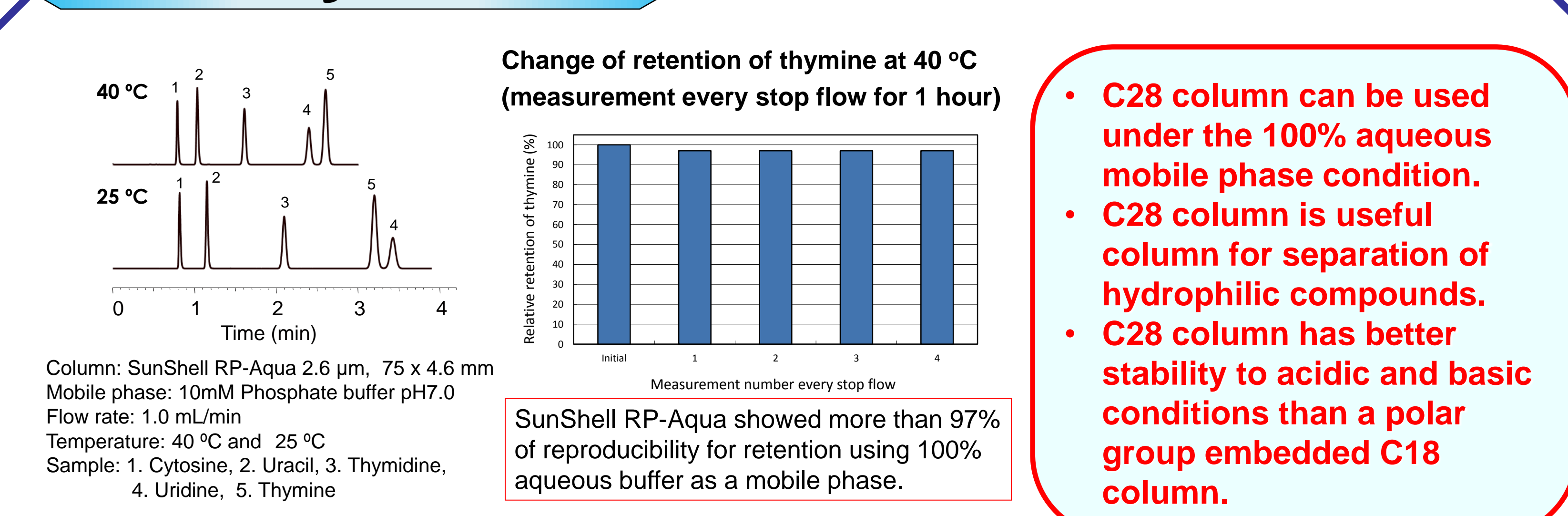
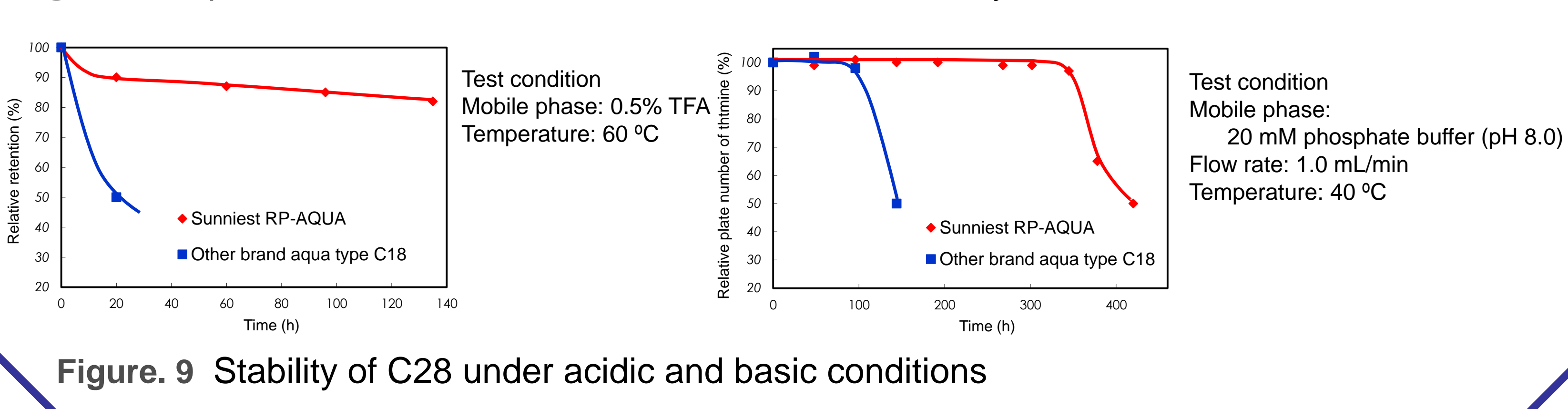
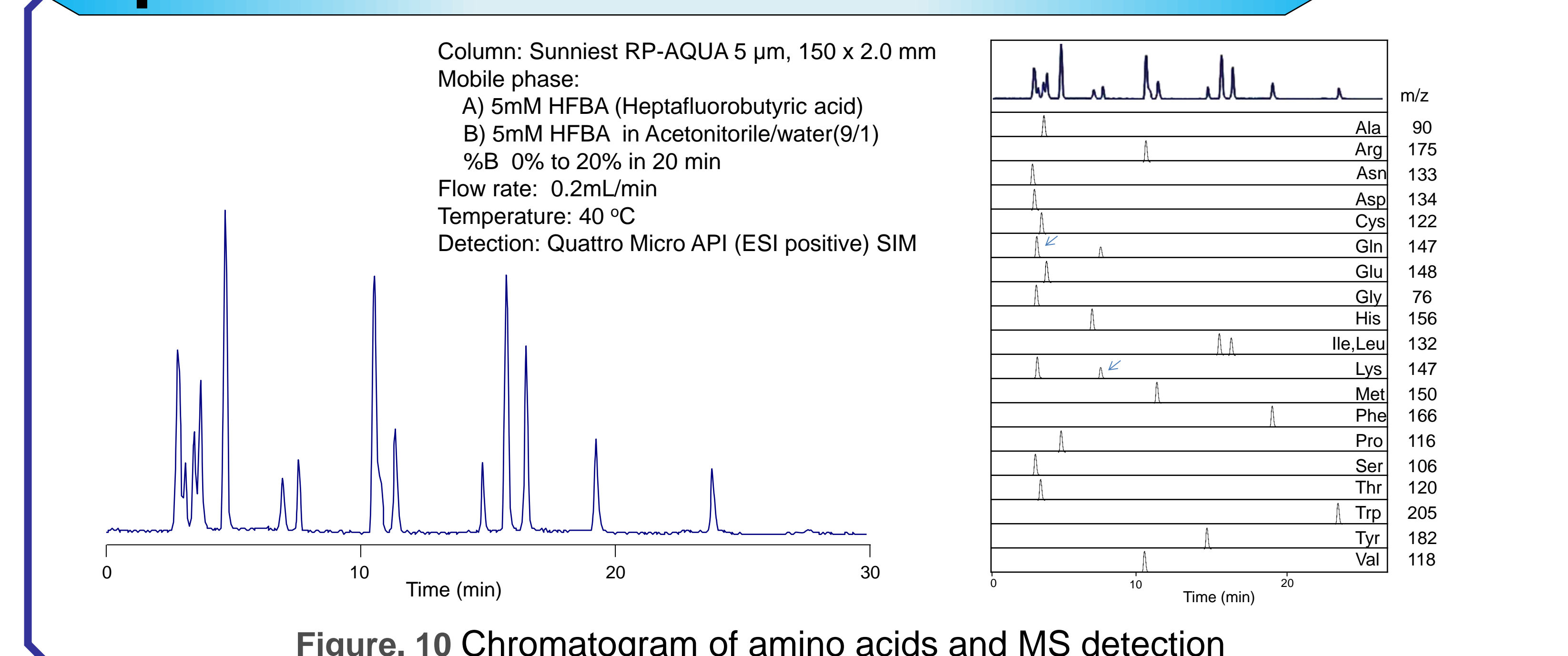


Figure. 8 Separation of nucleobases and retention time stability



Separation of amino acids with C28



SunShell



Sunrise

HPLC Column

ChromaNyk
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